From benzene make *m*-bromoaniline:

In this reaction three reactions are required.

1) A nitration

2) A conversion from the nitro group to an amine

3) A bromination

Because the end product is *meta* a *meta* directing group must be utilized. Of the nitro, bromine, and amine group, only the nitro group is *meta* direction. This means that the first step need to be the nitration and not the bromination. Also, the conversion of the nitro group to an amine must occur last because the amine group is *ortho/para* direction.

From benzene make *p*-nitropropylbenzene

In this reaction three reactions are required.

1) A Friedel Crafts acylation

2) A conversion from the acyl group to an alkane

3) A nitration

Because the propyl group has more than two carbons, it must be added in two steps. A Friedel Crafts acylation followed by a Clemmensen Reduction. Remembeer that Friedel Crafts reactions are hindered if the benzene ring is strongly deactivated. This means that the acyl group must go on first. Because the end product is *para* a *para* directing group must be utilized. Of the nitro, acyl, and alkane group, only the alkane group is *meta* direction. This means that the acyl group must be converted to an alkane prior to the nitration step.
Contributors

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